

# Chemical gloves

The devil is in the details.

**Daniel R. Kuespert, PhD, CSP**  
**Homewood Laboratory Safety Advocate**





# Recent incidents at JHU

- H<sub>2</sub>S exposure/Bloomberg
  - Materials synthesis
- Fire/AML/Wyman Park
  - Laser cutter/foamcore
- Exotic needlestick/Remsen
  - Needle blow-off
- Liquid N<sub>2</sub> exposure/Remsen
  - Replacing Dewar on Schlenk line trap
- Glassware serious injury/Croft
  - Catching a dropped beaker

- 1. What similarities do these incidents have?**
- 2. What might have prevented the incidents?**
- 3. What caused the incidents?**



# Safety classes at JHU

- Formal
  - EN.500.601 (1cr): *Research Laboratory Safety*
  - EN.540.490 (1cr): *Intro to Chemical Process Safety*
- Informal
  - Pretty much any topic
  - Currently developing *Cryogen Safety* seminar

# Chemical gloves

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# KIMBERLY-CLARK\* Nitrile Glove Chemical Resistance Guide



## The Science of Protection.

Use the color code rating system below with the chart at right to determine the chemical compatibility for incidental exposure.

### GREEN

The results for this specific chemical suggest that the glove would provide an adequate barrier for use in most applications.

A glove/chemical combination receives a **GREEN** rating if:

- The permeation breakthrough time is excellent or good and the chemical has high volatility.
- OR**
- The permeation breakthrough time is excellent and the chemical has low volatility.

### YELLOW

The results require additional consideration to determine suitability for use.

A glove/chemical combination receives a **YELLOW** rating if:

- Any glove/chemical combination does not meet either set of conditions required for a **GREEN** or **RED** rating.

### RED

Not recommended for use.

A glove/chemical combination receives a **RED** rating if:

- The permeation breakthrough time is poor and the chemical has low volatility.
- OR**
- The permeation breakthrough time is not recommended and the chemical has either high or low volatility.

Chemical Name	Permeation Time (minutes) ASTM F739-99A	Permeation Rate (pg/cm <sup>2</sup> /min) ASTM F739-99A	Concentration	Color Code Rating
Acetaldehyde	<1	353	99.5%	RED
Acetic Acid	5	482	99.7%	RED
Acetone	1	466	99.5%	RED
Acetonitrile	1	329	99%	RED
Acrylic Acid	1	57.8	99%	RED
Ammonium Hydroxide	7	395	30%	RED
Amyl Acetate	4	261	99%	RED
Aniline	7	74.7	99.5%	RED
Benzaldehyde	78	0.57	99.5%	YELLOW
Benzene	<1	627	99.8%	RED
Benzyl Alcohol	5	86.8	99%	RED
n-Butanol	10	5.99	99.8%	YELLOW
Butyl Acetate	3	233	99%	RED
Carbon Disulfide	2	3.81	99%	RED
Carbon Tetrachloride	5	48.9	99.5%	RED
Chloroform	1	958	99%	RED
Citric Acid	>480	Not Detected	50%	GREEN
Cyclohexane	>480	Not Detected	99.7%	GREEN
Cyclohexanol	112	1.18	99%	GREEN
Cyclohexanone	1	787	99.8%	RED
d-Limonene	107	0.157	97%	GREEN
n-Dibutyl Phthalate	>480	Not Detected	99%	GREEN
1,2-Dichlorobenzene	<1	1179	99%	RED
Dichloromethane	1	2006	99.9%	RED
Diesel Fuel, mixture	160	0.63	Mixture	GREEN
Diethyl Ether	1	595	99.9%	RED
Diethylamine	<1	587	99.5%	RED
Di-isobutyl Ketone	10	1141	80%	YELLOW
Dimethyl Sulfoxide	8	501	99.90%	RED
Dibutyl Phthalate	>480	Not Detected	99%	GREEN
1,4-Dioxane	<1	707	99.4%	RED
Ethanol	7	296	99.5+ %	RED
Ethanolamine	>480	Not Detected	99%	GREEN
Ethidium Bromide	90	0.68		GREEN
Ethylene Glycol	>480	Not Detected	99.8%	GREEN
Formaldehyde	110	0.172	37%	GREEN
Formic Acid	6	0.554	88%	RED
2-Furaldehyde	<1	385	99%	RED
Glutaraldehyde	>480	Not Detected	50%	GREEN
n-Hexane	16	55.3	99+ %	YELLOW
Hydrazine	21	10.2	98%	YELLOW

# What we'll cover

- Why it's not so simple—the complications
- How to determine which glove to use
- Concerns for which to watch out

Why choosing gloves  
is hard

## Chemical resistance of gloves depends on:

### Chemical properties

- Identity
- Concentration

### Glove properties

- Material
- Thickness
- Surface quality
- Overall quality

### Material properties

- Breakthrough time
- Permeation rate
- Physical resilience

### Mode of use

- Possible contact
- Mechanical abuse
- Motion/flexing

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Cyclohexanol	112	1.18	99%	

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# Chemical identity

- Manufacturers test a limited number of chemicals against their gloves.
- ChemWatch Gold SDSs have computer-predicted “glove selection index”.

## Recommended material(s)

### GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

**“Forsberg Clothing Performance Index”.**

The effect(s) of the following substance(s) are taken into account in the **computer-generated** selection:

ACETONITRILE

Material	CPI
BUTYL	A
BUTYL/NEOPRENE	A
CPE	A
PE/EVAL/PE	A
PVA	A
SARANEX-23	A
NEOPRENE	B
TEFLON	B
NATURAL RUBBER	C
NATURAL+NEOPRENE	C
NITRILE	C
VITON/NEOPRENE	C

\* CPI - Chemwatch Performance Index

A: Best Selection

B: Satisfactory; may degrade after 4 hours continuous immersion

C: Poor to Dangerous Choice for other than short term immersion

**NOTE:** As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as “feel” or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

# Concentration

- Not every concentration is tested.
- Some gloves good at low concentration fall apart at higher concentrations.

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# Glove material

- Many different materials used
- Same material may be different between manufacturers or even models of glove!
- **Do not trust** glove chemical resistance data for other gloves.



Lab Depot, Inc.

# Glove thickness

- Thicker material resists chemicals longer—usually.
- Thicker material means less dexterity.



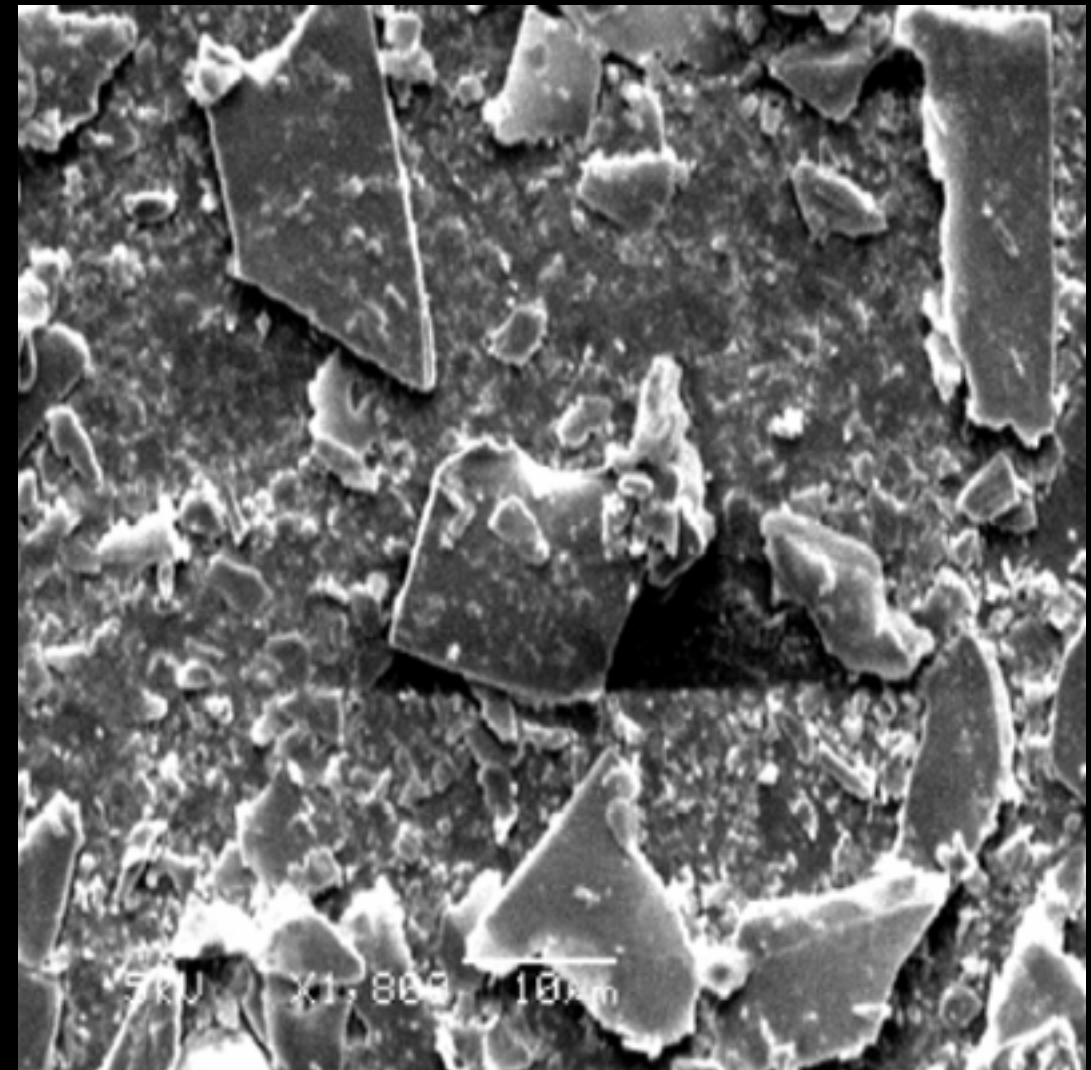
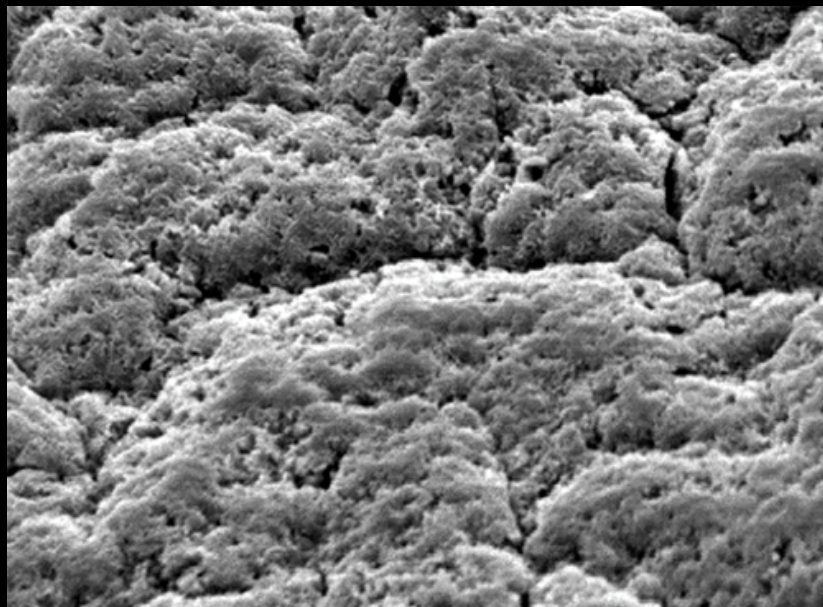
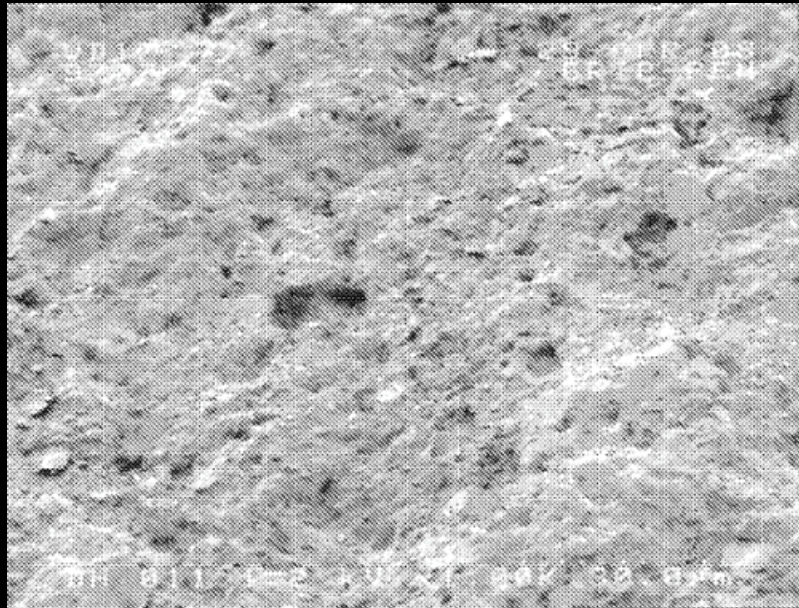
Dow DuPont



Lab Depot, Inc.



# Surface quality



# Glove quality

- Cheap gloves are cheap for a reason.

















# Breakthrough time

ALL CLEAR

ONLY SELECTED CHEMICALS

Chemical name	CAS number
<input checked="" type="checkbox"/> Battery Acid   70%	7664-93-9
<input checked="" type="checkbox"/> Battery Acid   47%	7664-93-9
<input checked="" type="checkbox"/> Battery Acid	7664-93-9
<input checked="" type="checkbox"/> Sulfuric Acid	7664-93-9
<input checked="" type="checkbox"/> Sulfuric Acid   70%	7664-93-9
<input checked="" type="checkbox"/> Sulfuric Acid   96%	7664-93-9
<input checked="" type="checkbox"/> Sulfuric Acid   47%	7664-93-9

						
707D	707FL	8500PF	8005	8005PF	8050PF	707HVO
						
Product URL	Product URL	Product URL	Product URL	Product URL	Product URL	Product URL
BDT	BDT	BDT	BDT	BDT	BDT	BDT
TTL INT	TTL INT	TTL INT	TTL INT	TTL INT	TTL INT	TTL INT
>240 NT	>240 NT	>240 >240	>240 >240	>240 >240	>240 >240	>120 NT
>480 NT	>480 NT	>480 >240	>480 >240	>480 >240	>480 >240	>480 NT
NT NT	NT NT	NT NT	NT NT	NT NT	NT NT	NT NT
NT NT	NT NT	NT NT	NT NT	NT NT	NT NT	NT NT
>240 NT	>240 NT	>240 >240	>240 >240	>240 >240	>240 >240	>120 NT
NT NT	NT NT	NT NT	NT NT	NT NT	NT NT	>10 NT
>480 NT	>480 NT	>480 >240	>480 >240	>480 >240	>480 >240	>480 NT

Breakthrough time is the amount of time it takes for the first molecule of chemical to penetrate the glove.



# Permeation rate

Chemical Name	Permeation Time (minutes) <i>ASTM F739-99A</i>	Permeation Rate (pg/cm <sup>2</sup> /min) <i>ASTM F739-99A</i>	Concentration	Color Code Rating
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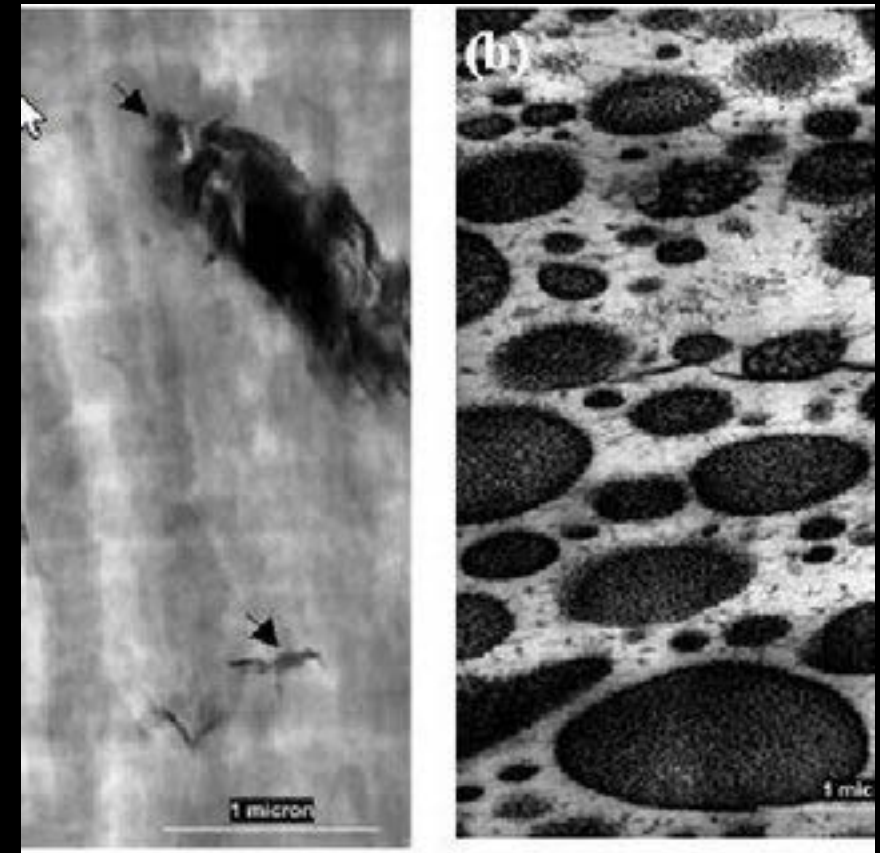
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Permeation rate is the steady-state rate at which chemical is transported through the glove material.



# Glove resilience

- Glove materials degrade with chemical contact.



Abd Razak, S.B. *J. Natural Sciences Research* 4(3) 2014.



# Possible contact

- Direct chemical immersion
- Intermittent chemical contact
- Chemical splash contact



# Mechanical abuse

- Sharps-handling can put invisible holes in gloves
- Wearing too small a size can stress the glove



# Motion/flexing

- Even just flexing your hand degrades the glove performance

What glove to use?



# Always

- Choose  $BT > 3 \times$  (the time you'll wear the glove)



# Direct chemical immersion

- No exam gloves
- Choose as long a BT as available
- Discard before BT reached



# Intermittent contact

- Choose BT based on work time
- Sometimes can use shorter BT if PR is low relative to chemical toxicity—ask an industrial hygienist





# Splash contact

- Very toxic substances
  - Double-glove
  - Engineer experiment to minimize contact
- Will you know you're splashed?
  - YES: Shorter BT is tolerable
  - NO:
    - Choose  $BT > 3 \times$  (work time)
    - Inspect gloves after use and discard if signs of contact

Special concerns

# Cheap gloves

- Often contain fillers like  $\text{CaCO}_3$
- These can improve properties in some cases
- Excess filler hastens degradation, though

# Latex gloves are not the only gloves that cause allergic reactions

- Carbamates and thiazoles used in latex, nitrile, vinyl can be allergens
- Anaphylactic shock reactions have occurred

# “Powder-free” gloves are not necessarily clean

- Processes to remove powder can leave residual chemicals

# DO NOT TRUST another manufacturer's data

- There are several ways to process a glove
- These produce different BT, PR, resilience
- Fillers differ, surface treatments differ, and base polymer differs

# BT/PR vary widely from lab to field

- Vinyl gloves often have so much plasticizer that they begin leaking immediately upon donning
- Quality issues like microscopic pores and surface imperfections in nitrile gloves are often not visible
- BT can drop 31% with movement like pipetting, so use a 3x safety factor

# What we covered

- Why it's not so simple—the complications
- How to determine which glove to use
- Concerns for which to watch out



# Thanks for listening!

I would be glad to take any questions now.




JOHNS HOPKINS  
UNIVERSITY



# Close calls

- “Almost-accidents:” no injury or property damage (including business interruption)
- Report close calls to [closecalls@jhu.edu](mailto:closecalls@jhu.edu)
- Close calls lead to incidents



# Tell your close-call story

- What happened?
- Was it really a close call or was it an incident?
- What could have happened?
- What were the causes?
- How could you prevent future occurrences?
- Were there higher-level things to address?